

SOLUTION STOICHIOMETRY

NAME _____

SECTION _____

- How many grams of sodium hydroxide are required to prepare 500.0 mL of 0.500 M NaOH?

 - Describe how you would prepare this solution.

- Calculate the volume of 0.750 M KNO_3 that contains 17.0 g of KNO_3 .

- What is the concentration of HCl in a 250.0 mL sample of hydrochloric acid if 15.5 mL of 0.0100 M NaOH is needed to react with all of the HCl?

